

## Synthesis, Characterization Of 1, 3-Thiazine Derivatives Using Chalcones, And Evaluation As Antimicrobial

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### Abstract

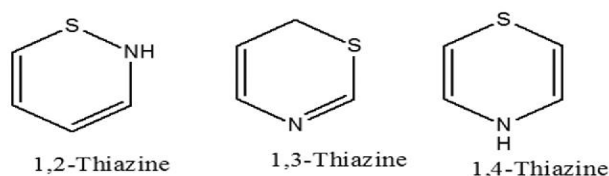
This study is concerned with the synthesis and characterization of the **1, 3-thiazine derivatives**. These compounds (**d-f**) were prepared by reacting of Thiosemicarbazide with the appropriate chalcones (**d-f**) in the presence of sodium hydroxide in ethanolrefluxed on water bath for 6 hrs, and the completion of the reaction was emphasize by thin layer chromatography (TLC), and the formed compounds were identified by NMR, IR,. The proposed reaction for their formations , And testing these organic compounds as antimicrobials

**Key word : Chalcone , NMR , IR ,thiazine .**

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### Introduction :

**Thiazine** is an organic compound containing a ring of four carbon, one nitrogen and one sulfur atom. There are three isomers of thiazine, 1,2-thiazine, 1,3-thiazine, and 1,4-thiazine, which differ by the arrangement of the nitrogen and sulfur atoms in the ring. Synthetic heterocyclic compounds especially containing heteroatoms like N, S, O have massive potential primarily as drugs<sup>1</sup>. Thiazine is a heterocyclic compound having one(N)and ( S ) atom at varied positions in the six membered ring exist as 1,2; 1,3; 1,4-thiazines and subsequently their derivatives<sup>2</sup>.



Structure of 1, 3- thiazines possesses an N-C-S linkage that is assent to be very useful units in the fields of medicinal and pharmaceutical chemistry. 1,3- thiazines and its derivatives have been reported to exhibit a variety of biological activities like Antibacteria<sup>3</sup> , Antifunga<sup>4</sup> , Antitubercular<sup>5</sup> , Antiinflammatory<sup>6</sup> , Analgesic<sup>7</sup> , Sedative-hypnotic<sup>8</sup> , Immunosuppressive agents<sup>9</sup> , Herbicida<sup>10</sup> .General synthesis involves synthesis of chalcones by ClaisenScmidth condensation reaction<sup>10,11</sup> followed by cyclo condensation of chalcones in the presence of thiosemicarbazide(N-Aminothiourea)giving 1, 3-Thiazines. The constitution of derivatives has been supported by elemental analysis, IR, NMR and Mass spectral data.

### **The Experimental**

All solvents were distilled / dried prior to use.All solvent were dried over anhydrous sodium sulphate unless other wise specified. <sup>13</sup>C NMR; <sup>1</sup>HNMR Spectroscopywere recorded using Bruker DRX system AL 500 (500 MHz). . IR spectra were recorded, usingshimadzu FT-IR affinity spectrophotometer in the Department of Chemistry, College of Science, Thi-Qar University, Iraq, as KBr disks. Only principal absorption bands of interest are reported and expressed in cm<sup>-1</sup>.

#### **1 : Prepare of Chalcones<sup>12,13</sup>**

Chalcones can be prepared by an aldol condensation between benzaldehyde and acetophenone in the presence of sodium hydroxide as a catalyst.

#### **1-Prepare of (E)-3-(4-bromophenyl)-1-(4-nitrophenyl)prop-2-en-1-one (a) :**

To equimolar mixture (0.01 moles) of 1-(4-nitrophenyl)ethanone and 4-bromobenzaldehyde in ethanol (50 ml), 2% NaOH solution (1 ml) was added and stirred for 10 hrs at room temperature and then refluxed for 6 hrs on water bath. The excess solvent was distilled off under vacuum and poured into ice cold water, filtered and recrystallized from ethanol.

#### **2 - Prepareof (E)-1-(4-(dimethylamino)phenyl)-3-(4-methoxyphenyl)prop-2-en-1-one (b) :**

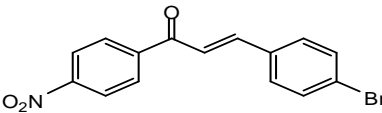
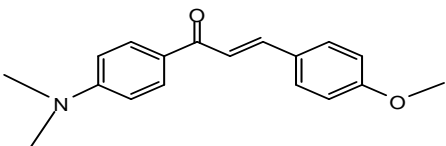
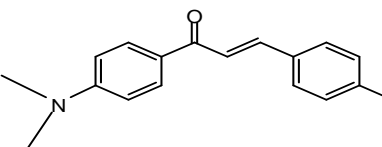
To equimolar mixture (0.01 moles) of 1-(4-(dimethylamino)phenyl)ethanone 4-methoxybenzaldehyde in ethanol (50 ml), 2% NaOH solution (1 ml) was added and stirred for 10 hrs at room temperature and then refluxed for 6 hrs on water bath. The excess solvent was distilled off under vacuum and poured into ice cold water, filtered and recrystallized from ethanol.

#### **3 - Prepareof (E)-3-(4-chlorophenyl)-1-(4-(dimethylamino)phenyl)prop-2-en-1-one (c) :**

To equimolar mixture (0.01 moles) of 1-(4-(dimethylamino)phenyl)ethanone4-chlorobenzaldehydein ethanol (50 ml), 2% NaOH solution (1 ml) was added and stirred for 10 hrs at room temperature and then

refluxed for 6 hrs on water bath. The excess solvent was distilled off under vacuum and poured into ice cold water, filtered and recrystallized from ethanol.

**Table(1)**

symbol	chalcone derivatives
<b>a</b>	
<b>b</b>	
<b>c</b>	

## 2 : Prepareof 1, 3-thiazine derivatives

### 1 - Prepareof 4-(4-bromophenyl)-2-hydrazinyl-6-(4-nitrophenyl)-6H-1,3-thiazine (d) :

To equimolar mixture (0.008 moles) of compound **(a)** and Thiosemicarbazide in ethanol (30 ml) was refluxed on water bath for 6 hrs. The reaction mixture was concentrated under vacuum, cooled and poured into ice cold water. The solid separated was filtered, dried and recrystallized from ethanol.

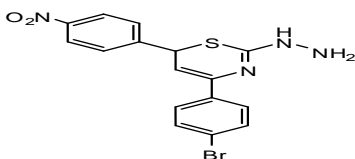
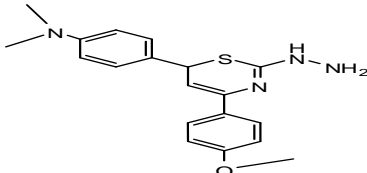
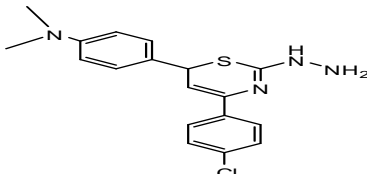
### 2 - Prepareof 4-(2-hydrazinyl-6-(4-methoxyphenyl)-6H-1,3-thiazin-4-yl)-N,N-dimethylaniline (e) :

To equimolar mixture (0.008 moles) of compound **(b)** and Thiosemicarbazide in ethanol (30 ml) was refluxed on water bath for 6 hrs. The reaction mixture was concentrated under vacuum, cooled and poured into ice cold water. The solid separated was filtered, dried and recrystallized from ethanol.

### 2 - 4-(6-(4-chlorophenyl)-2-hydrazinyl-6H-1,3-thiazin-4-yl)-N,N-dimethylaniline (f) :

To equimolar mixture (0.008 moles) of compound **(c)** and Thiosemicarbazide in ethanol (30 ml) was refluxed on water bath for 6 hrs. The reaction mixture was concentrated under vacuum, cooled and poured into ice cold water. The solid separated was filtered, dried and recrystallized from ethanol.

**Table(1)**

symbol	1, 3-thiazine derivatives
<b>d</b>	
<b>e</b>	
<b>f</b>	

#### Antimicrobial activity<sup>14,15</sup>

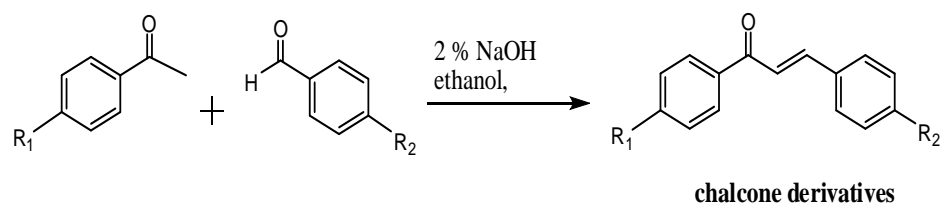
Different concentrations of the prepared compounds (0.1,0.2,0.3)mg/ ml were used to study their antimicrobial activity in vitro against two tested bacteria (Staphylococcus aureus)as gram positive and bacteria ( E. Coli) as gram negative bacteria. Also Antifungal activity was screened against Candida albicans in Muller Hinton agar medium,preparation of nutrient broth, dilution and application were carried out using the same procedure as for antimicrobial testing.The plates were incubated at 30 °C for 48 hours (Fungi spp.) or 37 °C for 24 hours (bacteria) and the antimicrobial activity was evaluated by measuring the diameter of the inhibition zone (IZ) around the disc in mm.

#### Determination of cytotoxicity using Human erythrocytes<sup>16</sup>

Different amounts of solutions of 1, 3-thiazine i.e., (0.2, 0.4, 0.6) mg/ml were prepared. serial dilutions of the compounds (d, e,f), were made in phosphate buffered saline. A total volume of 0.8 ml for each dilution was placed in an Eppendorf tube. A negative control tube (containing saline only) and a positive control tube (containing tap water) were also included in the analysis. human erythrocytes were added to each tube, to give a final volume of 1 ml. Solutions were incubated at 37°C for 30 min. The tubes were then examined for red blood cell decomposition.

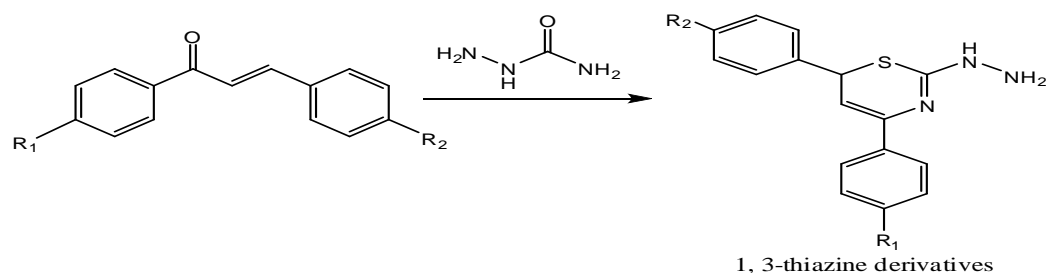
#### Results and discussion

In this paper, the **1, 3-thiazine** were prepared, where in the first step the chalcone derivatives were prepared from the reaction of equal quantities of aldehydes derivatives and acetophenone derivatives (ketone) by condensation aldol. The proposed reaction for their formations was shown as below in (scheme 1).



**scheme (1)**

1, 3-thiazine compounds were prepared from the reaction of chalcones with the semicarbazide compound, and the completion of the reaction was verified by thin layer chromatography (TLC), and the formed compounds were identified by NMR, IR, mass spectra. The proposed reaction for their formations was shown as below in (scheme 2).



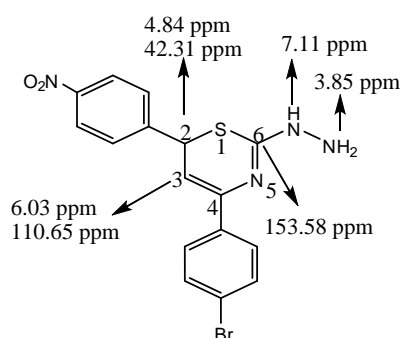
**(scheme 2)**

The structures of these Chalcones were established on the basis of spectral data, **IR**, spectra of these compounds (**a,b,c**) showed strong stretching absorption band at 1723, 1738 and 1748  $\text{cm}^{-1}$  ppm respectively for (**C=O**) as shown Figure(1-1), (1-2), (1-3), **IR**, spectra of these compounds (**a,b,c**) showed stretching absorption band at 1597, 1621 and 1629  $\text{cm}^{-1}$  respectively for (**C=C**) as shown Figure(1-1), (1-2), (1-3).

The structures of these 1,3 Thiazine were established on the basis of spectral data, **IR**,  **$^1\text{H}$ -NMR**,  **$^{13}\text{C}$  NMR**. **IR** spectra of these compounds (**d,e,f**) showed strong stretching absorption band at 1622, 1621 and 1615  $\text{cm}^{-1}$  ppm respectively for (**C=N**) as shown Figure(1-1), (1-2), (1-3), **IR** spectra of these compounds (**a,b,c**) showed stretching absorption band at (3430-3347), (3417-3334) and (3340-3281)  $\text{cm}^{-1}$  respectively for (**NH<sub>2</sub>**), also **IR** spectra of these compounds (**d,e,f**) showed stretching absorption band at 3239, 3221 and 3176  $\text{cm}^{-1}$  ppm respectively for (**NH**) as shown Figure(1-4), (1-5), (1-6).

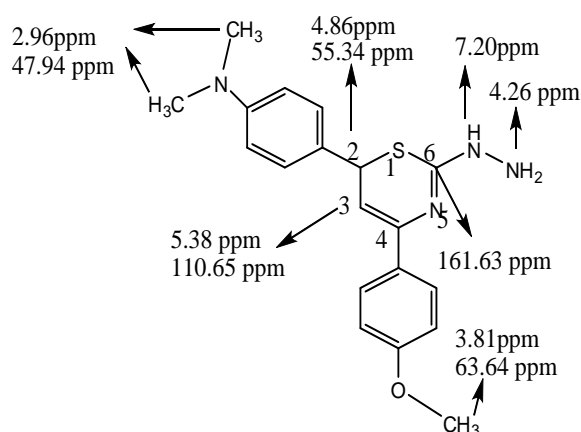
The  $^1\text{H-NMR}$  of **d** showed singlet peak  $\text{C}_2\text{-H}$   $\delta$ (4.48) and  $\text{C}_3\text{-H}$   $\delta$ (6.03 ppm). The  $^1\text{H-NMR}$  spectrum of (**d**) showed singlet peak  $\text{NH}_2$   $\delta$ (3.85 ppm) and  $\text{NH}$  at  $\delta$ (7.11 ppm). Finally,  $^1\text{H-NMR}$  spectrum of **d** showed aromatic protons integrated 8H at  $\delta$ (7.31- 8.22) ppm shown in Figure (1-1)

The  $^{13}\text{C NMR}$  spectrum of the **d** showed resonance between  $\delta$ 121.44-151.19 ppm which assigned to the carbon<sup>30,31,32</sup> group were belonged to the aromatic carbons. C-2 and C-3 were appeared at  $\delta$  42.31 and 110.65 ppm respectively, whereas the resonance at  $\delta$  153.58 ppm were assigned to the C-6 atom respectively figure (1-2).



The  $^1\text{H-NMR}$  of (**e**) showed singlet peak  $\text{C}_2\text{-H}$   $\delta$ (4.86) and  $\text{C}_3\text{-H}$   $\delta$ (5.83 ppm). The  $^1\text{H-NMR}$  spectrum of (**e**) showed singlet peak  $\text{NH}_2$   $\delta$ (4.26 ppm) and  $\text{NH}$  at  $\delta$ (7.20 ppm). The  $^1\text{H-NMR}$  spectrum of **e** showed singlet peak for methoxy group ( $\text{O-CH}_3$ ) at  $\delta$  3.81 ppm. The  $^1\text{H-NMR}$  spectrum of **d** showed singlet peak (equivalent protons) for two methyl groups at  $\delta$  3.204 ppm ( $\text{s, 6H, 2CH}_3\text{-N}$ ). Finally,  $^1\text{H-NMR}$  spectrum of **d** showed aromatic protons integrated 8H at  $\delta$ (6.77- 7.44) ppm shown in Figure (1-3)

The  $^{13}\text{C NMR}$  spectrum of the **d** showed resonance between  $\delta$ 112.98-153.44 ppm which assigned to the carbon group were belonged to the aromatic carbons. C-2 and C-3 were appeared at  $\delta$  55.34 and 110.56 ppm respectively, .O-C in methoxy group were appeared at  $\delta$ 63.46 ppm, , The  $^{13}\text{C NMR}$  spectrum of the **f** showed resonance singlet peak (equivalent carbons) for two methyl groups ( $\text{CH}_3$ )<sub>2</sub>-N at  $\delta$ 47.94 ppm, whereas the resonance at  $\delta$  161.36 ppm were assigned to the C-6 atom respectively figure (1-4).





		Gram positive bacteria staph.aureas		
<b>d</b>	0.1	14	12	7
	0.2	16	14	10
	0.3	19	17	11
<b>e</b>	0.1	18	19	10
	0.2	19	20	12
	0.3	21	22	14
<b>f</b>	0.1	13	11	9
	0.2	15	15	10
	0.3	18	17	11

#### Determination of cytotoxicity using Human erythrocytes

The results of cytotoxicity of the prepared compounds in the direction of human red blood cells showed that the compounds (d, e,f) do not carry any toxicity at the concentrations (0.2, 0.4, 0.6) mg/ml. Red blood cells have been used to detect the toxicity of prepared compounds because this method is inexpensive, easy to apply and quick results. Red blood cell decomposition depends on the concentration of the material, incubation period and temperature. The red blood cell crash is due to the breakdown of the red cell membrane due to the co-toxicity between the toxic substances and groups (1) found in the initial structure of the protein

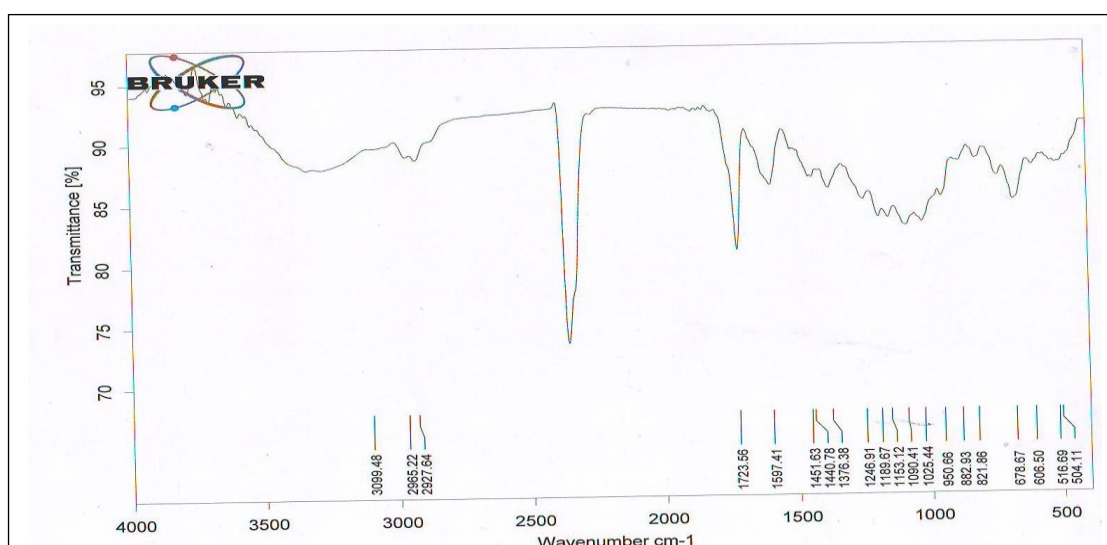




Figure (1-1) IR for (a)

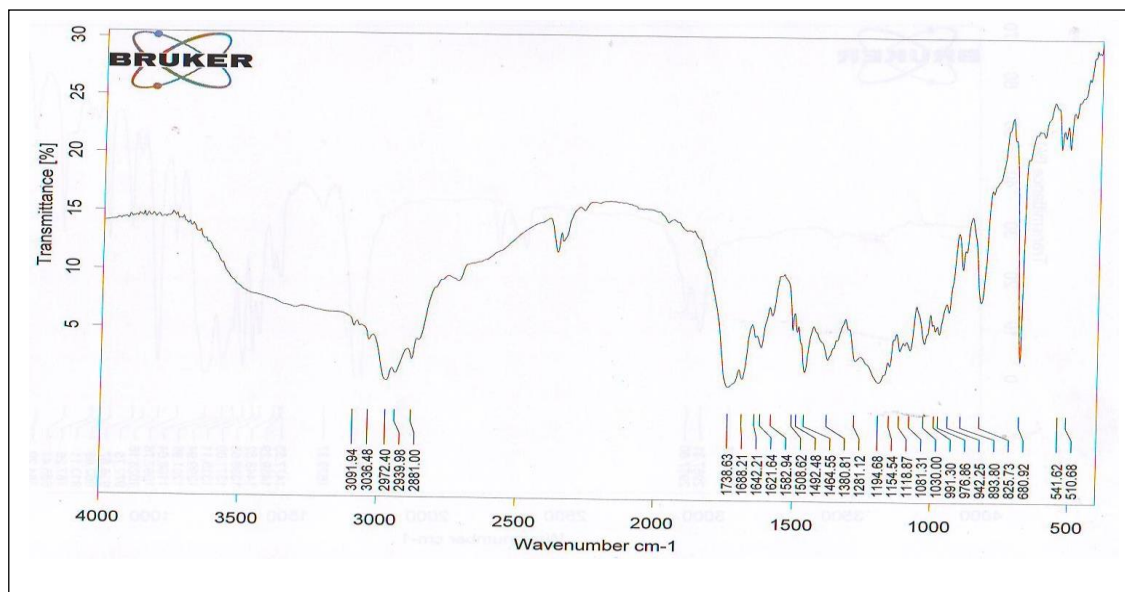


Figure (1-2) IR for (b)

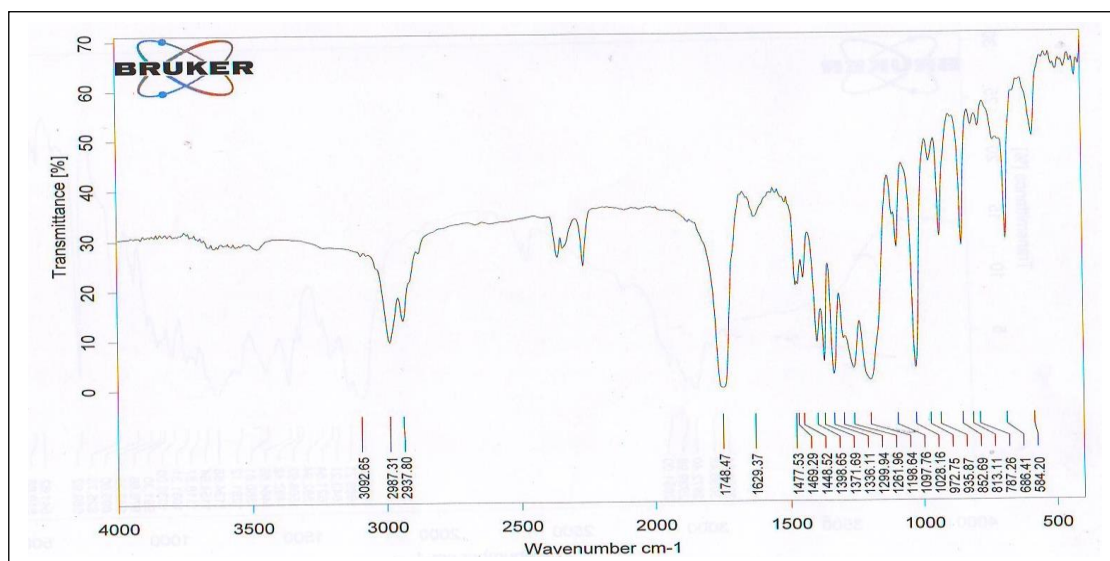


Figure (1-3) IR for (c)

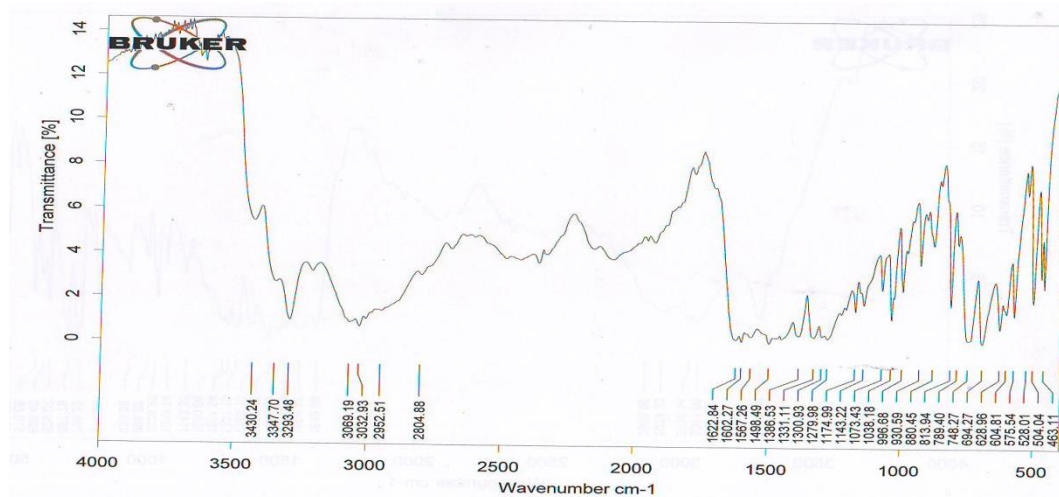


Figure (1-4) IR for (d)

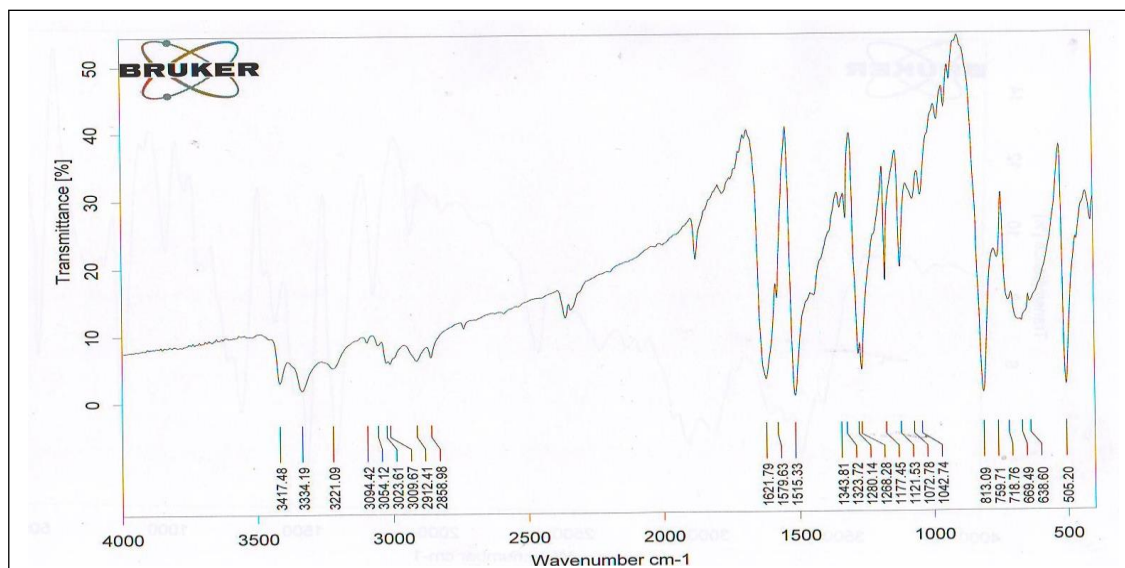


Figure (1-5) IR for (e)

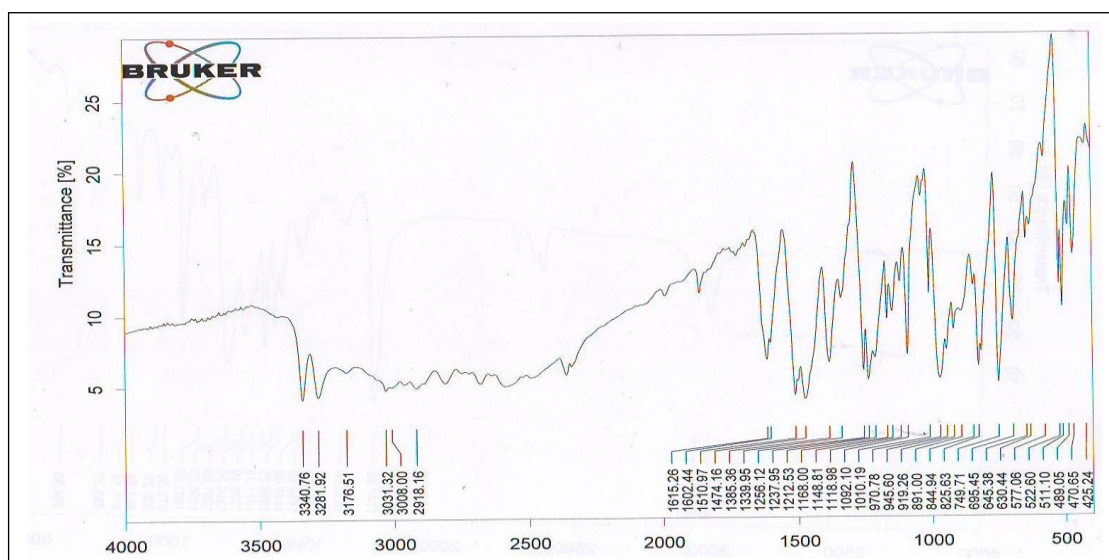


Figure (1-6) IR for (f)

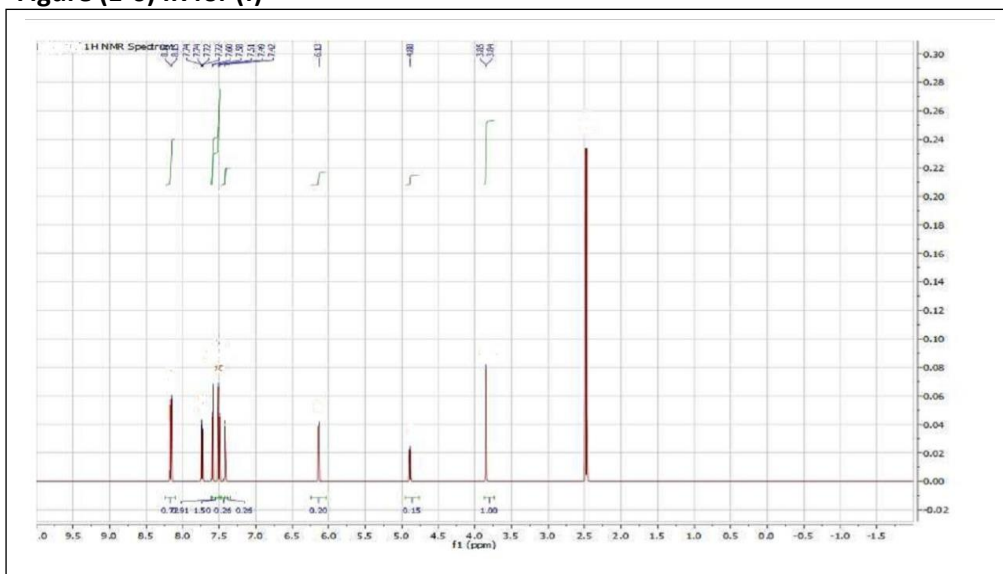


Figure (1-7) <sup>1</sup>H NMR of d

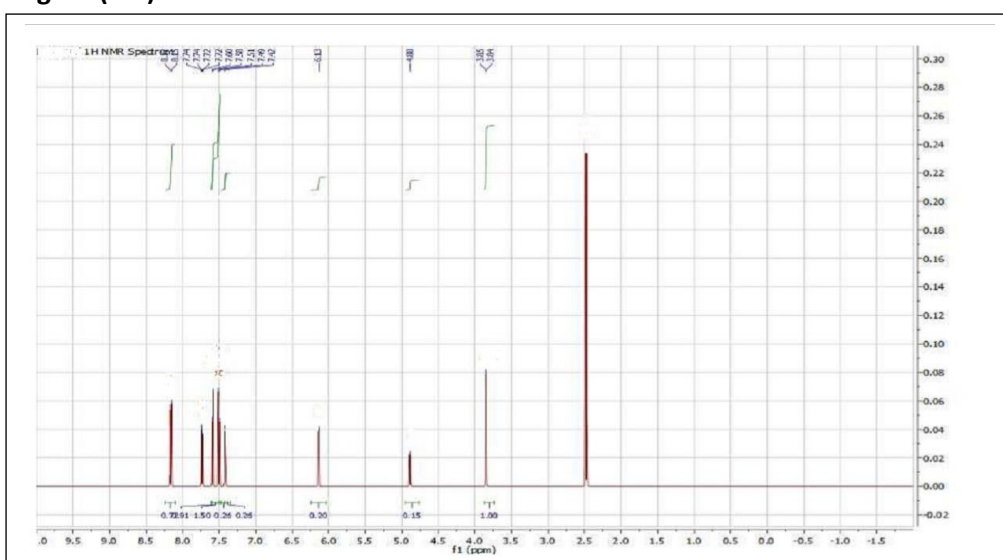
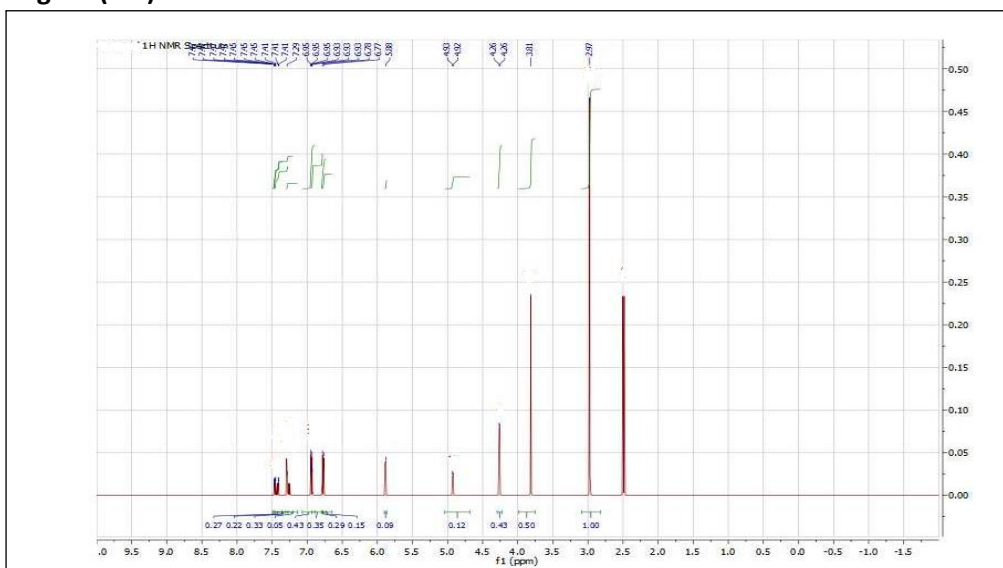
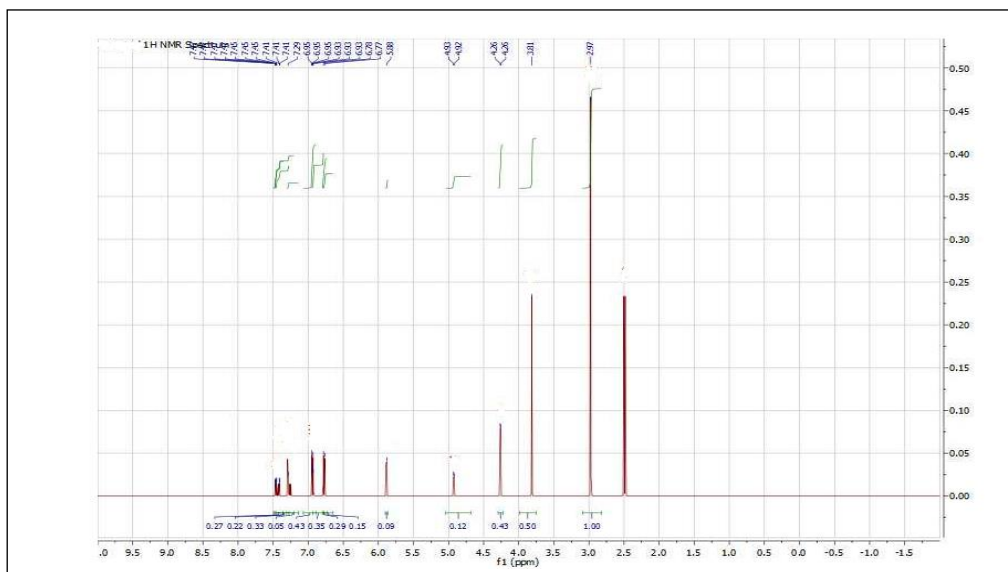


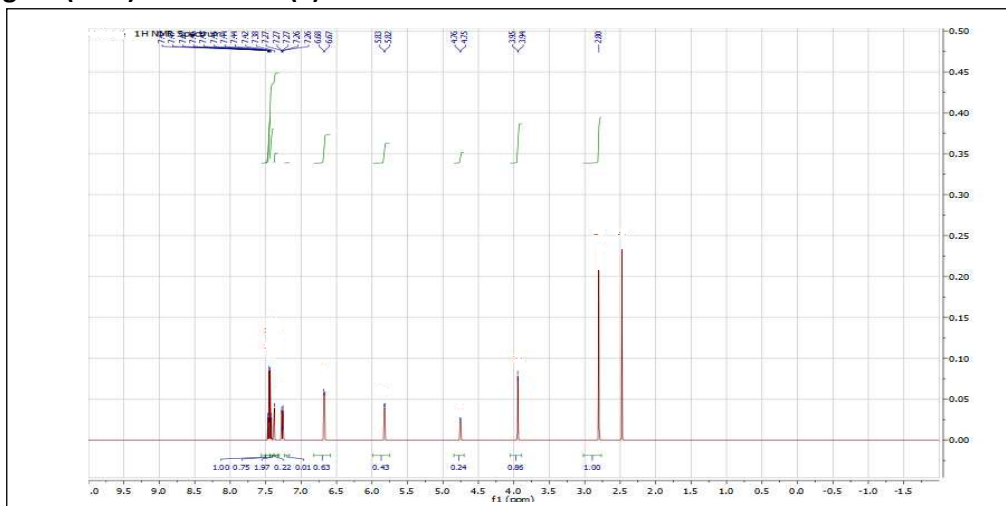
Figure (1-8) <sup>13</sup>C NMR of d



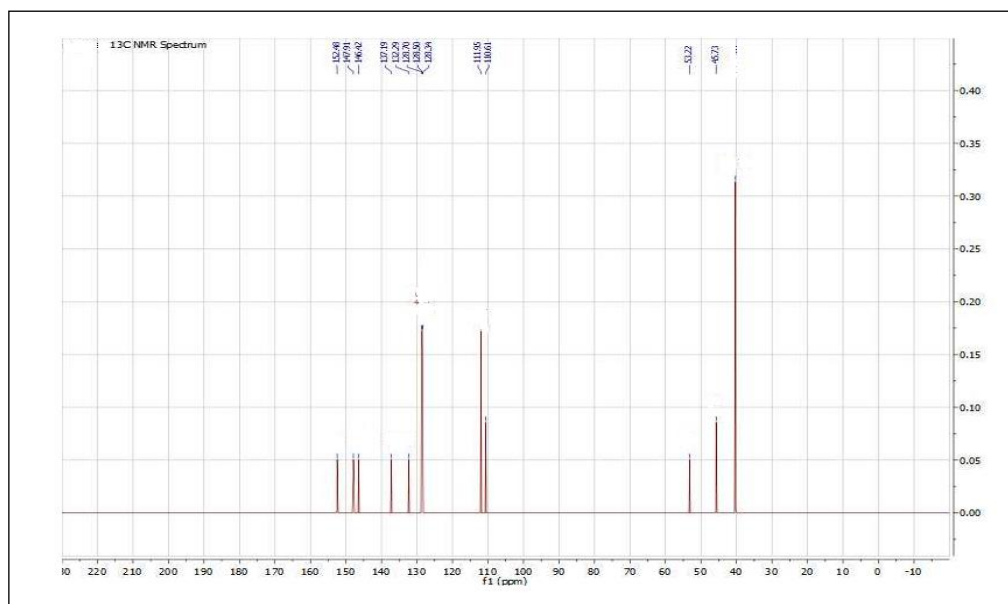
**Figure (1-9)  $^1\text{H}$ NMR of (e)**



**Figure (1-10)  $^{13}\text{C}$ NMR of (e)**



**Figure (1-11)  $^1\text{H}$ NMR of (f)**



**Figure (1-12) <sup>1</sup>H NMR of (f)**

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