

Assessment of Seasonal Variation in Water Quality and Water Quality Index of Tattekere Lake, Bangalore, India

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Abstract

Wetlands are transition zones on the planet that play an important role in nutrient dynamics and primary productivity. Bangalore and its urban agglomerations have many artificial wetlands developed for various hydrological purposes to meet the needs and water requirements of the city. The wetlands of Bangalore are subject to anthropogenic pressures, primarily due to the continuous discharge of wastewater, which has altered the chemical integrity of the wetlands. The current study was conducted to evaluate seasonal variations in physicochemical parameters of water quality and to calculate the water quality index of Tattekere Lake in Bangalore. Tattekere Lake, one of the largest lakes in Bangalore, Karnataka, India, originates in the central Gottigere area and traverses several major agricultural areas as well as various companies that receive enormous wastewater from various sources. Water samples were collected from four locations and analyzed for pH, electrical conductivity, TDS, dissolved oxygen, nitrate, BOD, total hardness, Ca, Mg, chlorides and alkalinity using standard methods. Seasonal variations in water quality measurements were recorded and compared to norms, and water pollution status was also examined.

Introduction

Currently, surface water pollution has received much attention globally. Natural process and anthropogenic activities, like hydrological features, climate change, precipitation, agricultural activities, and wastewater discharge from industries, are the main reason for worsening of surface water quality [1–3]. Nonetheless, rivers and/or lakes have also been used for cleaning and dumping purposes. This practices more prominent in developing countries, mostly in India, China, Africa etc. Rivers are one of the aquatic environments most prone to pollution due to their extensive flow that transports industrial, municipal, and agricultural wastes through the runoff [4]. Natural processes (precipitation and weathering) as well as anthropogenic activities such as domestic sewage, industrial pollution, and agricultural operations all have a significant impact on the quality of surface water in diverse locations. Domestic sewage and industrial wastewater discharge are point sources of pollution, but agricultural pollution is a nonpoint source of pollution that varies seasonally based on the climatic conditions of each region [5]. The pollutant concentration in river water varies seasonally due to variations in precipitation [6]. Nutrients in surface water have been linked mostly to land use activities [7]. As a result of anthropogenic activities, both point and nonpoint pollution sources enrich surface water with nutrients.

While animal manure and inorganic fertilizers used in agricultural fields are nonpoint sources of pollution that help to enrich the aquatic environment with nutrients, municipal wastewater and industrial wastewater are the main sources of nutrient pollution in the aquatic environment [8]. Due to their toxicity, persistence, and bioaccumulation effect on the ecosystem, heavy metal pollution of surface water is the primary concern [9]. A river can receive heavy metals from a variety of sources, both natural and manmade. The amount of heavy metals in rivers is typically negligible in clean areas and primarily results from

soil and rock weathering [10]. Raw industrial wastewater from factories, mining activities, sewage, and agrochemicals from agricultural areas are the main anthropogenic sources of heavy metal in rivers [10][11]. Tattekere Lake, one of the major lakes in Bangalore, Karnataka, India, which originates from the central Gottigere area, crosses different extended agricultural farms as well as various industries that receive large effluents from different sources. In the country, all of the prevailing industries and main town (Bangalore) with in the upper watershed have no proper treatment plants resulting in polluting the river [12,13]. Previously, there has not been any work on spatial and seasonal variation in physicochemical parameters and heavy metal in Tattekere Lake. The aim of this study was therefore to evaluate the level of different physicochemical parameters and heavy metals in terms of space and season in Tattekere Lake.

Materials and methods

Study area

The Tattekere Lake is the most important lake in Bangalore, India and serves as home to 9.2 million inhabitants. The lake rises on the central Gottigere area, Bangalore and crosses different extended agricultural farms as well as various industries that receive large effluents from different sources. The total length of the main course is some 50 km surrounding.

Water sampling

Sampling strategy was designed to cover a wide range of physiochemical parameters and heavy metals at sampling sites in Tattekere Lake. Water sampling was carried out on seasonal basis, namely during dry season (March–May, 2020) and rainy season (June–August, 2020). A total of 48 water samples were collected from eight sampling stations (24 samples during rainy season and 24 during dry season). Sampling, preservation, and transportation of the water samples to the laboratory were as per standard method [14].

Analysis of water samples

The samples were analyzed for 19 parameters, namely water temperature (WT), pH, electrical conductivity (EC), turbidity, nitrate nitrogen (NO₃-N), nitrite nitrogen (NO₂-N), ammonia nitrogen (NH₄-N), total nitrogen (TN), total phosphorus (TP), dissolved oxygen (DO), biological oxygen demand (BOD), chemical oxygen demand (COD), iron (Fe), zinc (Zn), copper (Cu), lead (Pb), chromium (Cr), Cadmium (Cd), and nickel (Ni). pH was measured on the sampling sites by pH meter model 370. WT and EC were also determined in the field using conductivity meter model CON 2700. All other parameters were determined in the laboratory following standard protocols [14]. TN (persulfate digestion method), NO₃-N (cadmium reduction method), NO₂-N (diazotization method), NH₃-N (Nessler method), TP (persulfate digestion followed by ascorbic acid method), COD (dichromate reactor digestion method) were analyzed by HACH DR/2400, whereas Fe, Zn, Cu, Pb, Cr, Cd, and Ni were analyzed using graphite atomic absorption spectrometer. Each analysis was performed in triplicate, and the mean value was taken. The analytical data quality was guaranteed through the implementation of laboratory quality assurance and quality control methods, including the use of standard operating procedures, calibration with standards, analysis of reagent blanks, recovery of known additions, and analysis of replicates.

Statistical analysis

Statistical analysis was performed by SPSS version 16.0 to calculate average mean, standard deviation, and Pearson's correlation (*r*) value to show the degree of physicochemical and metal association in river water. The ANOVA test (level of significance $\alpha = 0.05$) was employed to understand the spatial and seasonal variation in the physico-chemical and heavy metal concentrations.

Result and discussion

Seasonal and spatial variation in physicochemical parameters

The concentration of physicochemical parameters in dry and wet season of Tattekere Lake is shown in Tables 1 and 2. During the study period, water temperature in Tattekere Lake showed some seasonal variation and ranged from 19.1 to 23.6 °C. As expected, water temperature was the highest during dry seasons and the lowest during wet seasons. The highest average water temperature values were recorded

at site 7 during both dry season (23.01 °C) and wet season (21.9 °C). The reason might be there has been drinking water treatment plant at sampling station 7 so that the wastewater which drains from the treatment plant makes the river water temperature rise. There is no significant variation in water temperature among the sampling sites (p > 0.05), while there was a significant difference in seasonal mean concentration of water temperature (p < 0.05). The mean water temperature value (22.2 °C) in the present study was higher than the average value (16.7 °C) in Tinishu Akaki River, Ethiopia [15], but it was substantially lower than the mean water temperature value (25.65 °C) in Upper Awash River, Ethiopia [16]. Mean pH values at all sampling stations were slightly acidic to alkaline. The pH ranged from 6.08 to 8.47. Site 6 showed higher pH value (8.45) during the dry season. The lowest pH value (6.08) was found at site 7 in dry season. The lowest pH might be the sludge from drinking treatment plant mainly aluminum sulfate which lowers the pH of the river water. The deposition of sediment at Koka reservoir (site 6) is responsible for pH elevation. There is a significant variation in mean pH value among the sampling sites in Awash River (p < 0.05), while there was no seasonal significant difference in mean pH value in Awash River. The average pH value (7.23) in the present study is lower than the mean value (8.44) reported from Guder River, Ethiopia [17], and in Upper Awash River, Ethiopia (8.33), but higher than the mean pH (6.54) value of Buriganga River, Bangladesh [18], Iguedo River, Edo State, Nigeria (5.65) [19].

The turbidity values in Awash River varied from 29.27 to 159.51 NTU (Tables 1 and 2). The highest mean turbidity values (139.61 NTU) were found at site 2 during wet season because of surface runoff from nearest agricultural land, and the lowest average value (36.4 NTU) of turbidity was recorded at sampling site 6 during dry season. Higher values were recorded during the raining season as compared to the dry season. This could be attributed to run off water from the agricultural farm which carries suspended materials into the river. The soil around Koka area is bare and hence highly susceptible to erosion during rainy seasons. Sampling sites 2, 3, and 4 had higher turbidity levels than the rest of the sampling sites. There is a significant spatial and seasonal variation (p < 0.05) in average turbidity value among sampling sites (Table 3). The mean turbidity value in Awash River duringrainy season (121.06 NTU) was substantially higher than the value of turbidity (57 NTU) in Walgamo River, Ethiopia [20], in Gudbahi River, Eastern Tigray, Ethiopia (9.6 NTU) [21].

Table 1 Phys	icochemical	water quali	ty paramete	ers at differe	ent locations	of the Tatte	ekere Lake d	uring dry
				season				
Parameters	Sampling	Station						
	S1	S2	S3	S4	S5	S6	S7	S8
WT (°C)								
Mean	21.57	22.48	22.8	22.06	21.81	21.32	23.01	22.5
	(1.07)	(0.9)	(0.83)	(1.04)	(1.17)	(1.06)	(0.78)	(1.03)
Range	Range	21.45-	21.86-	20.96-	20.52-	20.19-	22.13-	21.40-
	20.40-	23.1	23.43	23.03	22.80	22.28	23.60	23.40
	22.50							
рН								
Mean	7.85	6.69	6.26	6.66	7.85	8.17	6.21	8.06
	(0.21)	(0.39)	(0.11)	(0.22)	(0.08)	(0.24)	(0.17)	(0.13)
Range	7.63–	6.28–	6.17–	6.41-	7.76–	7.99–	6.08–	7.92–
	8.04	7.07	6.38	6.82	7.93	8.45	6.41	8.17
EC (µS/cm)								
Mean	331.83	673.12	612.97	529.11	615.43	316.55	732.58	482.52
	(38.96)	(47.4)	(26.18)	(31.74)	(96.54)	(28.25)	(10.93)	(29.83)
Range	294.27-	626.68-	589.47-	498.21-	504.72-	286.17-	720.94–	449.07-
	372.06	721.43	641.19	561.63	682.14	342.02	742.62	506.38
Turbidity (NT	J)							
Mean	40.07	72.67	64.12	56.43	49.19	36.4	54.48	43.27
	(5.54)	(10.65)	(8.13)	(5.47)	(4.69)	(9.57)	(4.58)	(4.88)
Range	34.49-	63.28-	55.82-	50.62-	45.09-	29.27-	50.11-	38.51-
	45.57	84.25	72.06	61.49	54.3	47.28	59.25	48.27
NO3-N (mg l-	1)							

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Mean	0.8	13.33	27.87	12.5	14.71	2.31	1.86	1.36
	(0.25)	(0.96)	(0.86)	(0.66)	(1.14)	(0.3)	(0.11)	(0.13)
	0.54	12.20	27.40	11.00	42.47	4.00	4.74	4.22
Range	0.51-	12.30-	27.10-	11.80-	13.4/-	1.98-	1.74-	1.23-
NO2-N (mg l-	1)	14.20	20.00	15.10	13.71	2.30	1.90	1.49
Mean	0.24	0.61	0.90	0.26	0.52	0.21	0.29	0.31
	(0.08)	(0.02)	(0.02)	(0.03)	(0.06)	(0.04)	(0.06)	(0.04)
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Range	0.15-	0.59–	0.87–	0.23-	0.46-	0.18-	0.24–	0.27–
	0.31	0.63	0.92	0.28	0.58	0.25	0.35	0.36
TN (mg -1)								
Mean	2.28	39.63	83.43	79.40	50.23	8.22	2.90	3.57
	(0.35)	(2.1)	(1.02)	(0.9)	(2.15)	(1.64)	(0.51)	(1.24)
Range	2.01–	37.60-	82.52-	78.52–	48.62-	6.57–	2.36–	2.27–
/	2.68	41.80	84.53	80.32	52.68	9.86	3.37	4.75
TP (mg I–1)								
Mean	0.08	0.17	0.27	0.19	0.09	0.12	0.04	0.11
	(0.06)	(0.04)	(0.04)	(0.15)	(0.04)	(0.07)	(0.03)	(0.02)
Range	0.02-	0.13-	0.25-	0.02-	0.04-	0.05-	0.02-	0.08-
DO(mg = 1)	0.15	0.21	0.31	0.29	0.13	0.19	0.07	0.13
DO (IIIg I=1)		1			1	1	1	
Mean	7.47	5.15	4.51	3.62	6.83	7.03	6.29	7.58
	(0.89)	(1.27)	(1.37)	(0.91)	(0.51)	(0.93)	(1.24)	(1.25)
Danga	C 49	4.26	2 22	2.02	6.22	6.00	F 17	6.22
Range	0.40- 8 21	4.50-	5.52- 6.01	5.02- 4.67	0.52-	0.09- 7 Q/	7.62	0.25- 8 71
BOD (mg l-1)	0.21	0.02	0.01	4.07	7.54	7.54	7.02	0.71
Mean	16.22	41.35	59.23	80.32	38.52	27.13	17.53	19.62
	(2.42)	(3.34)	(0.94)	(3.64)	(0.88)	(4.81)	(3.25)	(1.82)
Range	13.69–	37.62-	58.26-	76.29–	37.58–	22.31–	13.85–	17.74–
	18.51	44.07	60.14	83.37	39.33	31.93	20.01	21.37
COD (mg l-1)								
Mean	27.33	72.63	147.98	112.3	53.24	40.5	125.0	35.55
	(4.45)	(10.41)	(2.77)	(1.32)	(1.72)	(3.39)	(1.11)	(1.09)
Range	22.85-	61.37-	144.95-	110.86-	51.40-	36.73-	123.77-	34.61-
	31.76	81.89	150.38	113.45	54.82	43.29	125.94	36.75

Table 2 Physi	Table 2 Physicochemical water quality parameters at different locations of the Tattekere Lake during wet										
season											
Parameters Sampling Station											
	S1	S2	S3	S4	S5	S6	S7	S8			
WT (°C)											
Mean	21.23	21 (0.72)	21.6	20.8	20.6	20.7	21.9	21.4			
	(1.17)	22.00	(1.06)	(1.18)	(1.08)	(1.44)	(0.66)	(0.6)			
Range	19.9–	20.2-	20.40-	19.50-	19.40-	19.10-	21.30-	20.80-			
	22.1 21.6 22.40 21.80 21.50 21.90 22.60 22.00										
рH											

Mean	8.13	6.55	6.71	6.64	7.55	7.73	6.27	8.0 (0.2)
	(0.20)	(0.18)	(0.1)	(0.06)	(0.33)	(0.26)	(0.19)	
	. ,	. ,	. ,	. ,	. ,	. ,	. ,	
Range	7.92-	6.39-	6.61-	6.60-	7.30-	7.51-	6.14-	7.79-
	8 31	6 74	6.81	6 71	7 92	8.02	6.48	8 19
FC (uS/cm)	0.01	•	0.01	0.72		0.01	0.10	0.120
Mean	285 5	589.6	521 73	476 47	279 97	294 53	648 27	577.6
Weath	(203.5	(10.78)	(25.15)	(12.60)	(18 / 5)	(20 10)	(12.25)	(17 71)
Banga	265 1	(19.78)	(25.15)	(12.05)	(18.45)	270.6	(13.23)	[17.71]
Kalige	205.4-	572.0-	490.8-	402.0- 407 E	201.7-	2/9.0-	661 2	505-
Turbidity (NITI	309.0	011.4	547.10	467.5	298.0	517.5	001.5	597.5
	J)	120.01	420.20	427.27	124.64	05.00	405.00	104.00
wean	122.8	139.61	138.26	137.37	124.64	95.08	105.83	104.89
	(12.31)	(21.02)	(10.56)	(20.04)	(14.91)	(6.85)	(16.07)	(14.97)
Range	110.6/-	117.62-	121.28-	115.38-	110.80-	89.17-	94.72-	92.07-
	135.28	159.51	154.36	154.60	140.42	102.59	124.26	121.35
NO3-N (mg I-	1)							
Mean	0.48	8.9	13.78	6.35	4.73	2.73	1.18	0.74
	(0.2)	(1.61)	(1.77)	(1.21)	(0.56)	(0.43)	(0.22)	(0.12)
Range	0.28-	7.14-	12.17-	5.11-	4.10-	2.25-	0.98-	0.63-
	0.68	10.3	15.68	7.53	5.17	3.07	1.42	0.86
NO2-N (mg l-	1)	2010	10:00	1.00	0.27	0.07		0.00
Mean	0.11	0.35	0.43	0.19	0.31	0 14	0.15	0.07
Wiedin	(0.03)	(0.07)	(0.06)	(0.02)	(0.03)	(0.05)	(0.04)	(0.01)
Pange	0.03	0.28_	0.27_	0.17_0.2	0.20		$0.11_0.2$	0.06-
Nalige	0.00-	0.28	0.37-	0.17-0.2	0.23-	0.09-	0.11-0.2	0.00-
$TN(mg \mid 1)$	0.15	0.42	0.49		0.34	0.10		0.07
110 (IIIg I-1)								
Mean	1.22	11.66	17.06	13.43	9.1	17.75	2.61	11.0
	(0.42)	(2.65)	(1.52)	(0.66)	(1.49)	(1.9)	(0.54)	(2.98)
Range	0.82-	9.17–	15.32-	12.86-	7.39–	16.14-	2.01–	8.25-
	1.65	14.45	18.16	14.15	10.15	19.85	3.07	14.16
TP (mg l-1)								
	0.05		0.45		0.47		0.07	
Mean	0.05	0.08	0.15	0.18	0.17	0.09	0.07	0.08
_	(0.03)	(0.05)	(0.08)	(0.11)	(0.08)	(0.06)	(0.05)	(0.04)
Range	0.03–	0.04–	0.06–	0.07–	0.09–	0.04–	0.03–	0.05–
	0.08	0.13	0.23	0.28	0.25	0.15	0.12	0.13
DO (mg l-1)								
Mean	10.82	4.60	4.25	5.12	6.24	6.41	7.27	8.62
	(2.46)	(1.46)	(1.02)	(1.22)	(2.55)	(1.19)	(1.98)	(1.71)
Range	8.69-	3.24-	3.27-	4.07-	4.48-	5.39-	5.81-	7.13-
hange	13 51	6.15	5 31	6.45	9 16	7 72	9 5 3	10.49
BOD (mg -1)	10.01	0.10	3.31	0.15	5.20	/.//2	5.55	10.15
505 (mg 1 1)								
Mean	11.13	14.43	34.09	38.32	17.49	12.81	16.63	13.24
	(1.92)	(2.89)	(1.2)	(1.5)	(0.81)	(1.79)	(1.65)	(1.97)
Range	9.14-	11.33-	32.84-	36.62-	16.92-	11.47–	15.19–	11.16-
	12.98	17.04	35.23	39.47	18.41	14.85	18.44	15.08
COD (mg l-1)			·		·			
Mag	10.00	40.0	04.4	C7 40	20.04	22.20	110.02	21.0
iviean	19.08	48.9	94.1	67.12	29.81	23.38	110.02	21.0
	(2.79)	(8.39)	(7.94)	(3.79)	(2.66)	(4.84)	(1./1)	(2.12)
капде	16.13-	41.9/-	88.4-	62.85-	27.14-	18.49-	108.93-	19.43-
1	21.68	58.23	103.2	70.09	32.47	28.16	112	23.42

Table 3 ANG	OVA relation	of physicoc	hemical para	ameters at d	ifferent sam	pling location	on and differ	rent season
Parameters	Dry seasor	า		Wet seaso	n		ANOVA	
	Mean (mg l–1)	Range	SD	Mean (mg l–1)	Range	SD	Spatial	Seasonal
WT	22.2	21.32– 23.01	0.6	21.15	20.6– 21.9	0.46	NS	SS*
рН	7.23	6.21– 8.17	0.84	7.2	6.27– 8.13	0.73	SS*	NS
EC	536.76	316.55– 732.58	152.33	459.21	279.97– 648.27	151.37	SS*	NS*
Turbidity	52.08	36.4– 72.67	12.36	121.06	95.08– 139.61 *	17.28	SS*	SS
NO3-N	9.34	0.8– 27.87	9.56	4.86	0.48– 13.78	4.66	SS*	NS
NO2-N *	0.42	0.21–0.9	0.24	0.22	0.07– 0.43	0.13	SS*	NS
NH4-N	0.78	0.12– 1.41	0.55	0.14	0.05– 0.29	0.07	SS*	SS
TN	33.71	2.28– 83.43	34.56	10.48	1.2– 17.75	6.05	SS*	NS
ТР	0.13	0.04– 0.27	0.07	0.11	0.06- 0.18	0.05	NS	NS
DO	6.25	4.51– 7.58	1.18	6.48	3.62– 10.82	2.41	SS*	NS
BOD	37.49	16.22- 80.32	22.68	19.77	11.13– 38.32	10.41	SS*	NS
COD	76.82	27.33– 147.98	45.81	51.68	19.08– 110.02	35.32	SS*	NS

The NO₃-N concentration varied from 0.28 to 28.8 mg l–1. The highest mean concentration (27.87 mg l⁻¹) of NO₃- N was found at site 3 during dry season because of intensive agricultural activities near to this site and animal manure waste near the river. The lowest average concentration (0.48 mg l–1) of NO₃- N was found at sampling site 1 during wet season. A significant variation in nitrate in the spatial trend was observed (p < 0.05). Nitrate is the most oxidized form of nitrogen found in aquatic environment, and during rainy season, considerable amount of nitrate washed from the agricultural farm and reached to water body through runoff.

The mean concentration of NO₃- N (9.34 mg l–1) in Awash River was higher than the average value (3.74 mg l⁻¹) from Jajrood River, Iran [22], from Vishwamitri River, India (0.06 mg l⁻¹) [22], from Sinos River, Brazil (0.3 mg l⁻¹) [23], but substantially lower than the average NO₃-N concentration (26.93 mg l⁻¹) from Chambal River, Rajasthan, India [24], from Mahanadi River, India (36.2 mg l⁻¹) [25], from Ogun River, Nigeria (35.18 mg l⁻¹) [26].

The NO₂- N concentration varied from 0.06 to 0.92 mg l⁻¹. The highest mean value (0.90 mg l⁻¹) of NO₂-N was reported at sampling site 3 during dry season, while the lowest mean concentration (0.07 mg l⁻¹) was observed at sampling site 8 during wet season.

The mean value (0.42 mg l⁻¹) of NO₂- N concentration in the present study was higher than the average value (0.06 mg l⁻¹) in Tigris River, Turkey [27], and also Elala River, Tigray, Ethiopia (0.11 mg l⁻¹) [28], while it is considerably lower than the average value (1.07 mg l⁻¹) in Awash River, Ethiopia [29].

The measured NH4-N values vary between 0.11 and 1.47 mg l–1 in dry season and between 0.03 and 0.35 mg l–1 in wet season. Site 4 showed higher average values (1.41 mg l⁻¹) during dry season while the lowest NH₄⁺ mean value (0.05 mg l⁻¹) was found at site 1 in wet season. There is a significant spatial and seasonal variation (p < 0.05) in mean NH₄- N values in Awash River (Table 3). NH4-N is a water-soluble gas that exists at low levels (0.1 mg l⁻¹) in natural waters. NH₄⁺ comes from the nitrogen-containing organic material and gas exchange between the water and the atmosphere [30]. It also derives from the biodegradation of waste and from domestic, agricultural, and industrial wastes.

The mean value (0.78 mg l⁻¹) of NH₄- N in Awash River was higher than the average value (0.07 mg l–1) from Upper Awash River, Ethiopia [16], Tigris River, Iraq (0.11 mg l⁻¹) [31]. The TN ranged from 0.82 to 84.53 mg l⁻¹ (Tables 1 and 2). The highest mean values (83.43 mg l–1) of TN have been noted at sampling site 3 in dry season, and the lowest average concentration (1.22 mg l⁻¹) was found at site 1 during wet season. There is a significant variation in mean TN values among sampling stations (p < 0.05); however, there was no seasonal significant difference in average TN concentration in Awash River.

The mean concentration (33.71 mg l^{-1}) of TN in the present study was very similar to the average TN (35.21 mg l^{-1}) in Walleme River, Ethiopia [32], but significantly higher than the mean TN value (2.06 mg l^{-1}) in Tigris River, Turkey (varol et al. 2011), from Xin'anjing River, China (1.55 mg l^{-1}) [33]. The concentration of TP varied from 0.02 to 0.31 mg l^{-1} in dry season and between 0.03 and 0.28 mg l^{-1} in wet season. Site 3 showed higher mean values (0.27 mg l^{-1}) during dry season while the lowest average TP value (0.04 mg l^{-1}) was found at site 7 in dry season. There was no a significant spatial and seasonal variation (p > 0.05) in average TP values in Awash River (Table 3).

The DO values varied from 3.02 to 13.51 mg l⁻¹. The DO was higher in wet season than in dry season at almost all sites. The low DO values in dry months were possibly due to considerable activities of microorganisms, which consumed appreciable amount of oxygen as a result of metabolizing activities and decay of organic matter. The highest mean values (10.82 51 mg l⁻¹) of DO were observed at site 1 during wet season. The lowest concentration (3.62 mg l⁻¹) of DO was found at site 4 during dry season, which receives agricultural runoff and animal manure wastes near the river. Dissolved oxygen is probably the most important parameter in natural surface water systems for determining the health of aquatic ecosystems [34].

The average value (6.48 mg l⁻¹) of DO in Awash River was very similar to the mean DO value (6.62 mg l⁻¹) from Blue Nile River, Ethiopia, but considerably higher than the mean DO value (1 mg l⁻¹) from Modjo River, Ethiopia, from Mahanadi River, India (4.58 mg l⁻¹) [25], from Ngong River, Kenya (4.35 mg l⁻¹). The concentration of BOD varied from 13.69 to 83.37 mg l–1 in dry season and between 9.14 and 39.47 mg l–1 in wet season. Site 4 showed higher average values (80.32 mg l–1) of BOD during dry season while the lowest average BOD value (11.13 mg l–1) was found at site 1 in wet season (Tables 1 and 2). There was a significant spatial variation (p < 0.05) in average BOD values in Awash River, whereas there was no significant seasonal variation (p > 0.05) in mean BOD values among the sampling sites (Table 3).

Based on the result of the present study, average BOD value (37.49 mg l^{-1}) was significantly higher than the mean value of BOD (24.23 mg l^{-1}) from Nyabugogo catchment, Rwanda [35], Gudbahri River, Eastern Tigray, Ethiopia (3.88 mg l^{-1}) [36], Rapti River, India (34.33 mg l^{-1}) [37], but lower than the mean value (38.10 mg l^{-1}) of BOD from Nile River, Egypt.

COD in Awash River varied from 16.13 to 150.38 mg l^{-1} . The highest average COD values (147.98 mg l^{-1}) were found at site 3 during dry season because of different agrochemicals' discharge to the river through runoff. The lowest mean value (19.08 mg l^{-1}) of COD was recorded at sampling site 1 during wet season. The average COD values were indicated a significant spatial variation (p < 0.05) among the sampling sites, but there was no seasonal variation in mean COD values in Awash River (Table 3). High values of COD indicate waterpollution, which is associated with wastewater discharged from industry or agricultural practices [38].

The mean value (76.82 mg l–1) of COD in Awash River was substantially lower than the average concentration (651 mg l–1) of COD from Modjo River, Ethiopia [39], from Buniganga River, Bangladesh [18].

The covariance matrix of the 12 analyzed variables was calculated from normalized data; consequently, it coincided with the correlation matrix (Tables 4 and 5). Because the eight sampling stations were combined to determine the correlation matrix, the correlation coefficients should be interpreted; however, they are affected simultaneously by spatial and seasonal variation.

There is a strong and positive correlation between (pH and EC, r = 0.805), (WT and BOD, r = 0.774), (NO3-N and NO2- N, r = 0.901), (NO₃-N and TN, r = 0.906), (NO₃-N and TP, 0.830), (NH₄-N and TN, r = 0.876), (NH₄-N and COD, r = 0.848), (TN and TP, r = 0.819), (TN and COD, r = 0.941). A significant negative correlation exists between (WT and turbidity, r = -0.812), (WT and DO r = -0.927), (TN and BOD, r = -0.854) during dry season (Table 4). Strong and positive correlations exist between (WT and BOD, r = 0.704), (turbidity and NO3-N, r = 0.749), (turbidity and NO₂- N, r = 0.722), (NO₃-N and NO₂-N, r = 0.921), (NO₃-N and BOD, 0.832), (TP and COD, r = 0.789). A significant negative correlation exists between (WT and NH₄-N, r = -0.769) during wet

season. The positive correlation probably indicated that these pollutants came from the same sources that are from agricultural runoff and animal manure.

	Table 4	4 Correlation r	natrix of	the	physico	chem	nical parame	eters du	ring	dry season					
	рН	WT	EC .	Tur	NO3	NO	2 NH4-	TN	ΤP	BOD	COD	D	С		
				bid	-N	-N	Ν								
			i	ity											
рН	1														
WT	- 0.7517	1													
EC	0.805307	- 0.75172	1												
Turbidity	0.649723	- 0.81201	0.78343	35	1										
NO3-N	0.311429	- 0.52466	0.45318	81	0.6872	205	1			r					
NO2-N	0.472314	- 0.4814	0.52556	67	0.7097	,	0.9007	06		1					1
NH4-N	- 0.1578	- 0.21762	0.20559	99	0.4449	8	0.76949	0.488	327	1					
TN	0.178033	- 0.51307	0.35822	13	0.6152	93	0.906192	0.645	128	0.8759	1				
										85					
ТР	0.205074	- 0.42186	0.10814	45	0.5697	'95	0.829653	0.698	926	0.6640	0.818	1			
										38	915				
BOD	- 0.35475	0.773919	- 0.449	93	- 0.75		-	-		-	-	-	1		
							0.69888	0.445	52	0.6826	0.853	0.			
										8	97	74			
												86			
COD	0.085421	- 0.48257	0.23312	19	0.5360	53	0.728213	0.391	149	0.8483	0.941	0.	0.	1	
										97	056	77	91		
												98	32		
												04	1		
DO	0.694409	- 0.92761	0.66614	49	0.6650	23	0.63082	0.518	756	0.3427	0.630	0.	-	0.	1
										35	267	52	0.	58	
												03	78	17	
												3	19	78	

	Table 5 Co	orrelation ma	atrix o	f the p	hysico	chemic	al paramet	ers du	iring v	vet seasor	ı				
	рН	WT	EC	Tur	NO3	NO2	NH4-	TN	TP	BOD	COD	D	0		
				bid	-N	-N	Ν								
				ity											
рН	1														
WT	- 0.7517	1													
EC	0.805307	-	1												
		0.75172													
Turbidity	0.649723	-	0.78	33435	1										
		0.81201													
NO3-N	0.311429	-	0.45	53181	0.6	6872	1								
		0.52466			05										
NO2-N	0.472314	- 0.4814	0.52	25567	0.7	'097	0.9007	06		1					
NH4-N	- 0.1578	-	0.20)5599	0.4	449	0.76949	0.48	383	1					
		0.21762			8			27							
TN	0.178033	-	0.35	58213	0.6	5152	0.906192	0.64	451	0.8759	1				
		0.51307			93			28		85					
ТР	0.205074	-	0.10)8145	0.5	697	0.829653	0.69	989	0.6640	0.818	1			
		0.42186			95			26		38	915				
BOD	- 0.35475	0.773919	- 0.	4493	- 0).75	-	-		-	-	-	1		
							0.69888	0.44	155	0.6826	0.853	0.			
								2		8	97	74			
												86			1
COD	0.085421	-	0.23	33119	0.5	360	0.728213	0.39	911	0.8483	0.941	0.	0.	1	
		0.48257			53			49		97	056	77	91		İ
												98	32		İ

									04	1		
DO	0.694409	-	0.666149	0.6650	0.63082	0.5187	0.3427	0.630	0.	I	0.	1
		0.92761		23		56	35	267	52	0.	58	
									03	78	17	
									3	19	78	

	Tak	ole 6 Mean cor	ncentratior	of heavy i	metals duri	ng dry seas	son				
Sites	Values	Metal Conce	Concentrations (mg I–1)								
		Fe	Zn	Cu	Pb	Cr	Cd	Ni			
Site-1	Mean	1.11 ± 0.58	0.74 ±	0.92 ±	0.56 ±	0.36 ±	0.07 ±	0.05 ±			
			0.62	0.62	0.17	0.12	0.02	0.01			
	Range	0.49–1.64	0.35-	0.29-	0.38-	0.26-	0.05-	0.03-			
			1.46	1.52	0.71	0.49	0.09	0.06			
Site-2	Mean	2.17 ± 0.8	1.12 ±	1.22 ±	0.70 ±	0.52 ±	0.09 ±	0.08 ±			
			0.78	0.83	0.21	0.2	0.36	0.01			
	Range	1.42-3.01	0.42-	0.41-	0.52-	0.32-	0.06-	0.06-			
			1.96	2.07	0.93	0.71	0.13	0.09			
Site-3	Mean	2.34 ± 0.92	1.42 ±	0.88 ±	0.84 ±	0.56 ±	0.13 ±	0.11 ±			
			1.21	0.47	0.43	0.09	0.05	0.04			
	Range	1.63–3.38	0.37–	0.35–	0.58–	0.46-	0.09–	0.07–			
			2.74	1.27	1.34	0.63	0.18	0.15			
Site-4	Mean	2.6 ± 1.0	1.22 ±	1.69 ±	0.77 ±	0.99 ±	0.18 ±	0.14 ±			
			1.10	0.96	0.61	0.31	0.07	0.06			
	Range	1.7–3.68	0.23-	0.73–	0.33-	0.65-	0.11-	0.08-			
			2.41	2.65	1.46	1.25	0.25	0.19			
Site-5	Mean	2.73 ± 1.03	1.56 ±	1.63 ±	1.36 ±	1.16 ±	0.22 ±	0.12 ±			
			1.27	1.19	1.20	0.35	0.07	0.03			
	Range	1.85–3.87	0.47–	0.49–	0.58-	0.77–	0.16-	0.09–			
			2.95	2.86	0.75	1.43	0.3	0.14			

	Table 6 Mean concentration of heavy metals during wet season ites Values Metal Concentrations (mg I=1)												
Sites	Values	Metal Conce	ntrations (mg -1)									
		Fe	Zn	Cu	Pb	Cr	Cd	Ni					
Site-1	Mean	1.11 ± 0.58	0.74 ±	0.92 ±	0.56 ±	0.36 ±	0.07 ±	0.05 ±					
			0.62	0.62	0.17	0.12	0.02	0.01					
	Range	0.49–1.64	0.35-	0.29–	0.38–	0.26-	0.05-	0.03–					
			1.46	1.52	0.71	0.49	0.09	0.06					
Site-2	Mean	2.17 ± 0.8	1.12 ±	1.22 ±	0.70 ±	0.52 ±	0.09 ±	0.08 ±					
			0.78	0.83	0.21	0.2	0.36	0.01					
	Range	1.42-3.01	0.42-	0.41-	0.52-	0.32-	0.06-	0.06-					
			1.96	2.07	0.93	0.71	0.13	0.09					
Site-3	Mean	2.34 ± 0.92	1.42 ±	0.88 ±	0.84 ±	0.56 ±	0.13 ±	0.11 ±					
			1.21	0.47	0.43	0.09	0.05	0.04					
	Range	1.63–3.38	0.37–	0.35–	0.58–	0.46-	0.09–	0.07–					
			2.74	1.27	1.34	0.63	0.18	0.15					
Site-4	Mean	2.6 ± 1.0	1.22 ±	1.69 ±	0.77 ±	0.99 ±	0.18 ±	0.14 ±					
			1.10	0.96	0.61	0.31	0.07	0.06					
	Range	1.7-3.68	0.23-	0.73–	0.33-	0.65-	0.11-	0.08-					

			2.41		2.65		1.46		1.25		0.25		0.19	
Site-5	Mean	2.73 ± 1.03	1.56	±	1.63	±	1.36	±	1.16	±	0.22	±	0.12	±
			1.27		1.19		1.20		0.35		0.07		0.03	
	Range	1.85-3.87	0.47-		0.49-		0.58-		0.77-		0.16-		0.09-	
			2.95		2.86		0.75		1.43		0.3		0.14	

Table 8	Table 8 ANOVA relation of heavy metals at different sampling location and different season Elements Dry season										
Elements	Dry season			Wet seasor	ı		ANOVA				
	Mean (mg I ⁻¹)	Range	SD	Mean (mg l–1)	Range	SD	Spatial	Seasonal			
Fe	2.17	1.11– 2.73	0.61	3.36	1.82– 4.12	0.77	NS	SS*			
Zn	0.64	0.46– 0.91	0.15	1.14	0.74– 1.56	0.3	NS	SS*			
Cu	0.7	0.44– 1.01	0.20	1.20	0.82– 1.69	0.34	NS	SS*			
Pb	0.59	0.31– 0.83	0.18	0.81	0.41– 1.36	0.29	NS	NS			
Cr	0.75	0.36– 1.16	0.29	0.63	0.3–0.98	0.25	SS*	NS			
Cd	0.06	0.03– 0.11	0.03	0.13	0.05– 0.24	0.07	SS*	SS*			
Ni	0.05	0.02– 0.09	0.02	0.10	0.03–0.2	0.07	SS*	SS*			

NS not statistical significant, SS statistical significant *p < 0.05



Fig. 1 Heavy metal concentration during dry and wet season

Conclusion

There is a significant spatial and seasonal variation in most of the physicochemical parameters in Tattekere Lake. The concentration of heavy metals during dry season is higher than the wet season except for Fe in which the highest concentration was found during wet season. Buffer zones should be protected in order to control soil and agricultural nutrients from entering to Tattekere Lake. Moreover, industries at the upper stream area should be properly and adequately treat the wastewater before discharging to the Tattekere

Lake and environmental protection agency need to regularly monitor and test the waste water based on the standard guidelines.

References

- [1] T.R. Green, M. Taniguchi, H. Kooi, J.J. Gurdak, D.M. Allen, K.M. Hiscock, H. Treidel, A. Aureli, Beneath the surface of global change: Impacts of climate change on groundwater, J. Hydrol. 405 (2011) 532– 560.
- [2] S.S.D. Foster, P.J. Chilton, Groundwater: the processes and global significance of aquifer degradation, Philos. Trans. R. Soc. London. Ser. B Biol. Sci. 358 (2003) 1957–1972.
- [3] N. Mujere, W. Moyce, Climate change impacts on surface water quality, in: Hydrol. Water Resour. Manag. Break. Res. Pract., IGI Global, 2018: pp. 97–115.
- [4] L. Nizzetto, G. Bussi, M.N. Futter, D. Butterfield, P.G. Whitehead, A theoretical assessment of microplastic transport in river catchments and their retention by soils and river sediments, Environ. Sci. Process. Impacts. 18 (2016) 1050–1059.
- [5] K.P. Singh, A. Malik, D. Mohan, S. Sinha, Multivariate statistical techniques for the evaluation of spatial and temporal variations in water quality of Gomti River (India)—a case study, Water Res. 38 (2004) 3980–3992.
- [6] M. Vega, R. Pardo, E. Barrado, L. Debán, Assessment of seasonal and polluting effects on the quality of river water by exploratory data analysis, Water Res. 32 (1998) 3581–3592.
- [7] M.N. Beaulac, K.H. Reckhow, AN EXAMINATION OF LAND USE-NUTRIENT EXPORT RELATIONSHIPS 1, JAWRA J. Am. Water Resour. Assoc. 18 (1982) 1013–1024.
- [8] D.G. Capone, R.P. Kiene, Comparison of microbial dynamics in marine and freshwater sediments: Contrasts in anaerobic carbon catabolism 1, Limnol. Oceanogr. 33 (1988) 725–749.
- [9] H. Ali, E. Khan, I. Ilahi, Environmental chemistry and ecotoxicology of hazardous heavy metals: environmental persistence, toxicity, and bioaccumulation, J. Chem. 2019 (2019).
- [10] R. Reza, G. Singh, Heavy metal contamination and its indexing approach for river water, Int. J. Environ. Sci. Technol. 7 (2010) 785–792.
- [11] J. Nouri, A.H. Mahvi, G.R. Jahed, A.A. Babaei, Regional distribution pattern of groundwater heavy metals resulting from agricultural activities, Environ. Geol. 55 (2008) 1337–1343.
- [12] H.S. Sudhira, T. V Ramachandra, M.H.B. Subrahmanya, Bangalore, Cities. 24 (2007) 379–390.
- [13] T. V Ramachandra, P.P. Mujumdar, Urban floods: Case study of Bangalore, Disaster Dev. 3 (2009) 1–98.
- [14] R.W. Bachmann, D.E. Canfield, Use of an alternative method for monitoring total nitrogen concentrations in Florida lakes, Hydrobiologia. 323 (1996) 1–8.
- [15] S. Melaku, T. Wondimu, R. Dams, L. Moens, Pollution status of Tinishu Akaki River and its tributaries (Ethiopia) evaluated using physico-chemical parameters, major ions, and nutrients, Bull. Chem. Soc. Ethiop. 21 (2007).
- [16] F.D. Fasil Degefu, A.L. Aschalew Lakew, Y.T. Yared Tigabu, K.T. Kibru Teshome, The water quality degradation of upper Awash River, Ethiopia., (2013).
- [17] B.W. Olbasa, Characterization of Physico-chemical Water Quality Parameters of River Gudar, Oromia region, West Shewa Zone, Ethiopia for Drinking Purpose, J. Appl. Chem. 10 (2017) 47–52.
- [18] S.S. Ahammed, S. Tasfina, K.A. Rabbani, M.A. Khaleque, An investigation into the water quality of Buriganga-a river running through Dhaka, Int. J. Sci. Technol. Res. 5 (2016) 36–41.
- [19] O.O. Udebuana, C.K. Akaluka, K.M.I. Bashir, Assessment of physico-chemical parameters and water quality of surface water of Iguedo River, Ovia South-West Local Government, Edo State, Assessment. 4 (2014).
- [20] B. Dessalew, T. Belina, D. Getachew, Assessment on the current water quality status of Walgamo River, Addis Ababa, Ethiopia, Int J Innov Res Sci Eng Technol. 6 (2017) 1–12.
- [21] M.M. Weldemariam, Physico-chemical analysis of Gudbahri river water of Wukro, Eastern Tigrai, Ethiopia, Int. J. Sci. Res. Publ. 3 (2013) 1–4.
- [22] H. Razmkhah, A. Abrishamchi, A. Torkian, Evaluation of spatial and temporal variation in water quality by pattern recognition techniques: A case study on Jajrood River (Tehran, Iran), J. Environ. Manage. 91 (2010) 852–860.
- [23] C. Steffens, C.R. Klauck, T. Benvenuti, L.B. Silva, M.A.S. Rodrigues, Water quality assessment of the

Sinos River–RS, Brazil, Brazilian J. Biol. 75 (2015) 62–67.

- [24] N. Gupta, S.M. Nafees, M.K. Jain, S. Kalpana, Physico-chemical assessment of water quality of river Chambal in Kota City area of Rajasthan State (India), Rasayan J. Chem. 4 (2011) 686–692.
- [25] S. Rout, A.K. Behera, A. Patnaik, Water Quality Analysis of River Mahanadi in Sambalpur City, Int J Sci Res Publ. 6 (2016) 266–270.
- [26] B.O. Dimowo, Assessment of some physico-chemical parameters of River Ogun (Abeokuta, Ogun State, Southwestern Nigeria) in comparison with national and international standards, (2013).
- [27] M. Varol, B. Gökot, A. Bekleyen, B. Şen, Water quality assessment and apportionment of pollution sources of Tigris River (Turkey) using multivariate statistical techniques—a case study, River Res. Appl. 28 (2012) 1428–1438.
- [28] F.G. Ftsum Gebreyohannes, A.G. Abraha Gebrekidan, A.H. Amanual Hadera, S.E. Samuael Estifanos, Investigations of physico-chemical parameters and its pollution implications of Elala River, Mekelle, Tigray, Ethiopia., (2015).
- [29] A.S. Keraga, Z. Kiflie, A.N. Engida, Evaluating water quality of Awash River using water quality index, Int. J. Water Resour. Environ. Eng. 9 (2017) 243–253.
- [30] D. V Chapman, Water quality assessments: a guide to the use of biota, sediments and water in environmental monitoring, CRC Press, 2021.
- [31] J. Abbas, Assessment of water quality in Tigris River-Iraq by using GIS mapping, Nat. Resour. 2013 (2013).
- [32] T. Minuta, D. Jini, Impact of effluents from wet coffee processing plants on the Walleme river of Southern Ethiopia, Res. J. Environ. Toxicol. 11 (2017) 90–96.
- [33] X. Li, P. Li, D. Wang, Y. Wang, Assessment of temporal and spatial variations in water quality using multivariate statistical methods: A case study of the Xin'anjiang River, China, Front. Environ. Sci. Eng. 8 (2014) 895–904.
- [34] H. Yang, Z. Shen, J. Zhang, W. Wang, Water quality characteristics along the course of the Huangpu River (China), J. Environ. Sci. 19 (2007) 1193–1198.
- [35] I. Nhapi, U.G. Wali, B.K. Uwonkunda, H. Nsengimana, N. Banadda, R. Kimwaga, Assessment of water pollution levels in the Nyabugogo Catchment, Rwanda, Open Environ. Eng. J. 4 (2011).
- [36] M. Mehari, Physico-chemical analysis of Gudbahri River Water of Wukro, Eastern Tigray, Ethiopia, Int J Sci Res Publ. 3 (2013) 1–4.
- [37] N.K. Chaurasia, R.K. Tiwari, Effect of industrial effluents and wastes on physico-chemical parameters of river Rapti., Adv. Appl. Sci. Res. 2 (2011) 207–211.
- [38] D. Bellos, T. Sawidis, Chemical pollution monitoring of the river pinios (Thessalia—Greece), J. Environ. Manage. 76 (2005) 282–292.
- [39] A. Mulu, T. Ayenew, S. Berhe, Impact of slaughterhouses effluent on water quality of Modjo and Akaki River in Central Ethiopia, Int. J. Sci. Res. 4 (2013) 899–907.